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The number of protons in the nucleus determines the atom's atomic number and its position in the Periodic Table. See more at atom.

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An atom is the smallest constituent unit of ordinary matter that has the properties of a chemical element. Every solid, liquid, gas, and plasma is made up of neutral

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- viniti

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Rutherford and the atomic nucleus

New Zealander Ernest Rutherford made waves in the scientific community in the first couple decades of the twentieth century with his work on the atomic structure.

Atomic nucleus - wikipedia, the free encyclopedia

The nucleus is the small, dense region consisting of protons and neutrons at the center of an atom. The atomic nucleus was discovered in 1911 by Ernest Rutherford

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